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Energy band alignment in MoS₂/HfO₂: Transfer-related artifacts and interfacial effects



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ABSTRACT

The knowledge of energy band alignment in heterojunctions with atomically thin transition metal dichalcogenides (TMDs) is critical for their use in advanced electronic and optoelectronic devices. Despite considerable efforts, the measurement of energy band offset across heterojunctions has been challenging, especially for van der Waals bonded stacks. Key obstacles are related to the scarce and often inconsistent information regarding the bandgap of the TMD layer and the offset between the conduction and valence bands, which is usually inferred from different measurement techniques and samples. To overcome this obstacle, we report combined internal photoemission (IPE) and photoconductivity measurements from 3-monolayer (ML) MoS₂ films, grown by chemical vapor deposition on sapphire and transferred onto HfO₂-covered silicon. We compare the spectral threshold of electron IPE in this heterostructure with IPE data from the Si/HfO₂ interface, yielding the value of the electrostatic potential variation. To improve band offset predictions, we examine the applicability of the classical electron affinity rule by deriving characteristic energies. Our results show that electronic properties at 2D TMD/insulator interfaces depend on the interface processing prior to the 2D material transfer, allowing for the modification of band offsets by adjusting the interface. Furthermore, the measured photoconductivity spectra of 3ML MoS₂ allow us to evaluate the bandgap of the TMD layer, which, combined with the IPE barriers, establishes the interface band diagram of a heterojunction. The presented IPE-based experimental approach can be extended to other two-dimensional TMDs for determining the corresponding band alignment schemes. It evaluates the impact of processing, such as solvent-based MoS₂ transfer, which introduces a dipole and alters band alignment.

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I. INTRODUCTION

The emergence of two-dimensional (2D) semiconductors has opened new avenues for next-generation electronic and optoelectronic devices, offering a compelling alternative to conventional silicon technologies.^{1,2} With their atomically thin structure, these materials exhibit exceptional electronic and optical properties, making them prime candidates for field-effect transistors,^{3–6} spintronic devices,^{7,8} and sensor applications.⁹ Notably, transition metal dichalcogenides (TMDs) such as MoS₂, MoSe₂, WS₂, and WSe₂ stand out owing to their intrinsic semiconducting properties,^{10,11} well-defined bandgap that can be tuned by changing the number of layers^{10,12} and high charge-carrier mobility.^{13,14} A key

advantage of these materials lies in their chemically inert surfaces,^{15,16} which facilitate ultra-fast charge transport in few-monolayer (ML) thickness range compared to traditional semiconductors such as silicon¹⁷ and indium gallium zinc oxide.¹⁸

Despite their promising properties, the integration of 2D materials with insulating layers faces critical challenges due to the screening and Coulomb scattering of charge carriers arising from impurities and defects at the interface. For example, empirical studies have demonstrated that amorphous dielectric substrates, such as SiO₂, can significantly reduce charge-carrier mobility in TMDs due to enhanced scattering.^{19–23}

Conversely, high-k dielectrics^{21,24} and atomically flat 2D insulators like hexagonal boron nitride (hBN) have been shown to

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mitigate these effects, leading to superior electrical performance. However, assessing these performance metrics often requires device fabrication, where processing steps can introduce additional nonidealities, complicating direct electrical characterization.^{25–27}

An important issue associated with the characterization of TMDs is related to the energy band alignment. In heterojunctions of 2D materials with 3D insulating oxides, electronic transport is influenced by the weak van der Waals (vdW) bonding between them. Crucial parameters that need to be known, and eventually controlled, to enable the fabrication of low-voltage devices, are the energy band offsets for electrons and holes. These offsets are determined by the relative alignment of the energy bands.

Energy band alignment at the interface between 2D materials and insulators is particularly intricate due to the unique nature of these weakly bonded heterostructures. Unlike conventional covalently bonded metal–oxide–semiconductor (MOS) systems, where defect states and interface trap densities can be well characterized, 2D materials introduce additional complexities. For example, SiO₂ is known to host defect bands at 2.6 and 4.6 eV below the conduction band (CB),²⁸ while HfO₂ possesses defect states approximately 2 eV below the CB.²⁹ In the case of 2D materials, interfacial species, such as water and hydrocarbons introduced during material transfer or exfoliation, can further influence energy band alignment.^{30,31} Additionally, polar hydroxyl (OH) groups on oxide surfaces have been linked to dipole formation and midgap trap states, as observed in transferred MoS₂ layers.^{31,32} Unlike traditional high-temperature SiO₂ growth on silicon, which forms a robust covalently bonded interface free of residual contaminants, the van der Waals interaction between 2D materials and insulators inherently permits the persistence of interfacial species, compromising interface passivation and stability.³³

From a technological standpoint, the use of multilayer 2D semiconductors instead of monolayers (MLs) can offer advantages in mitigating some of these interface-related challenges. Thicker layers exhibit reduced susceptibility to ambient degradation,³⁴ lower contact resistance due to a smaller bandgap,^{35,36} and enhanced current densities. In this study, we investigate the ballistic transport properties of photo-excited electrons across multilayer MoS₂ interfaces with two technologically relevant dielectrics, SiO₂ and HfO₂. Furthermore, we propose a methodology to qualitatively assess band discontinuities at these interfaces, providing insight into optimizing 2D material integration with insulating substrates for device applications.

II. EXPERIMENTAL DETAILS

Large-area (cm² scale) 3ML MoS₂ films were synthesized via DC magnetron sputtering of Mo in a sulfur vapor ambient³⁷ on boron-doped (10¹⁵ cm⁻³) (100)-oriented silicon, on which 25 nm oxide was grown by thermal oxidation. The quality of these films, assessed by x-ray photoelectron spectroscopy, Raman scattering, atomic force microscopy (AFM), and other methods, is detailed elsewhere.^{38,39}

As an alternative to direct growth on SiO₂/Si, a second series of MoS₂ films were synthesized on c-plane sapphire by metalorganic chemical vapor deposition (MOCVD) with Mo(CO)₆, H₂S, H₂, and N₂ precursors at 850 °C and then transferred to HfO₂/TiN/

SiO₂/p-Si substrates via a two-step water intercalation process.⁴⁰ The average thickness of the MoS₂ film was 2.5 nm that corresponds to three monolayers, as determined by Rutherford backscattering spectrometry and AFM.⁴¹ The target substrate comprised 20 nm of TiN, formed by physical vapor deposition onto 50 nm of SiO₂ that was thermally grown on p-type Si(100). TiN acted as a buried metal gate, decoupling the high-k HfO₂ (12 nm) grown by atomic layer deposition (ALD) using HfCl₄ and water precursors from the Si/SiO₂ substrate. Before MoS₂ transfer, the HfO₂ surface was annealed in a forming gas (95:5, N₂:H₂) at 450 °C for 15 min. For internal photoemission (IPE) measurements, both MOS capacitor structures [Fig. 1(a)] had ~100 nm thick Au pads that are known to form reliable electrical contacts with MoS₂,^{42,43} while a blanket Al layer on the silicon wafer's backside served as the back contact.

The Au pads were thermally evaporated through a shadow mask to prevent MoS₂ damage. This measurement geometry allows charge carrier generation studies in monolayer semiconductors beneath 10–20 nm thick semitransparent metal electrodes.^{43,44} It further enables detecting the photogeneration of charge carriers in discontinuous semiconductor layers, which is otherwise impossible.

The quantum yield $Y(h\nu)$ of IPE was defined as photocurrent normalized to the incident photon flux and analyzed as a function of photon energy. Photocurrent was measured at room temperature by applying a gate bias V_g and illuminating the sample with monochromatic light from a 150 W Xe arc lamp and Czerny-Turner monochromator. The photocurrent was recorded with a Keithley 6517 A electrometer. The use of Si and MoS₂ electrodes in the same capacitor allows one to use the Si(100) valence band (VB) top edge as reliable energy reference level, enabling the evaluation of conduction band positions in SiO₂, Al₂O₃,^{43,45} and HfO₂ insulators with an accuracy of approximately 50–100 meV. The stability of the Si VB across interface conditions stems from minimal dipole or charge effects.^{45,46}

To accurately estimate the voltage drop across the insulating oxides, the flatband voltage, corresponding to zero strength of the electric field in the oxide insulator, is determined from IPE current–voltage measurements under illumination with a specific photon energy, at which the photocurrent changes sign. The applied bias, corrected for the flatband voltage, is then divided by the oxide thickness to calculate the average electric field strength. For the 3ML MoS₂/HfO₂/TiN/SiO₂/p-Si stack, the flatband voltage was measured as $V_{fb} = -0.45 \pm 0.1$ V, while for the 3ML MoS₂/SiO₂/p-Si stack structure, it was $V_{fb} = -0.2 \pm 0.1$ V.

III. RESULTS AND DISCUSSION

In MOS capacitors, a 2D TMD film can function as both a light absorber and photo-electron emitter.^{12,42,47} Figure 2(a) shows the photocurrent spectrum, plotted in the Powell coordinates $Y^{1/3}(h\nu)$,⁴⁸ measured at -1.5 V for the 3ML MoS₂/HfO₂/TiN/SiO₂/p-Si structure. The spectrum features three distinct regions corresponding to different electronic excitation processes. The first region, spanning 1.5–3 eV, is attributed to transient photoconductivity, including the generation of A and B excitons⁴⁹ [see the inset in Fig. 2(a)], dissociation, trapping, and carrier collection. Transient photoconductivity is linked to trap states, as evidenced

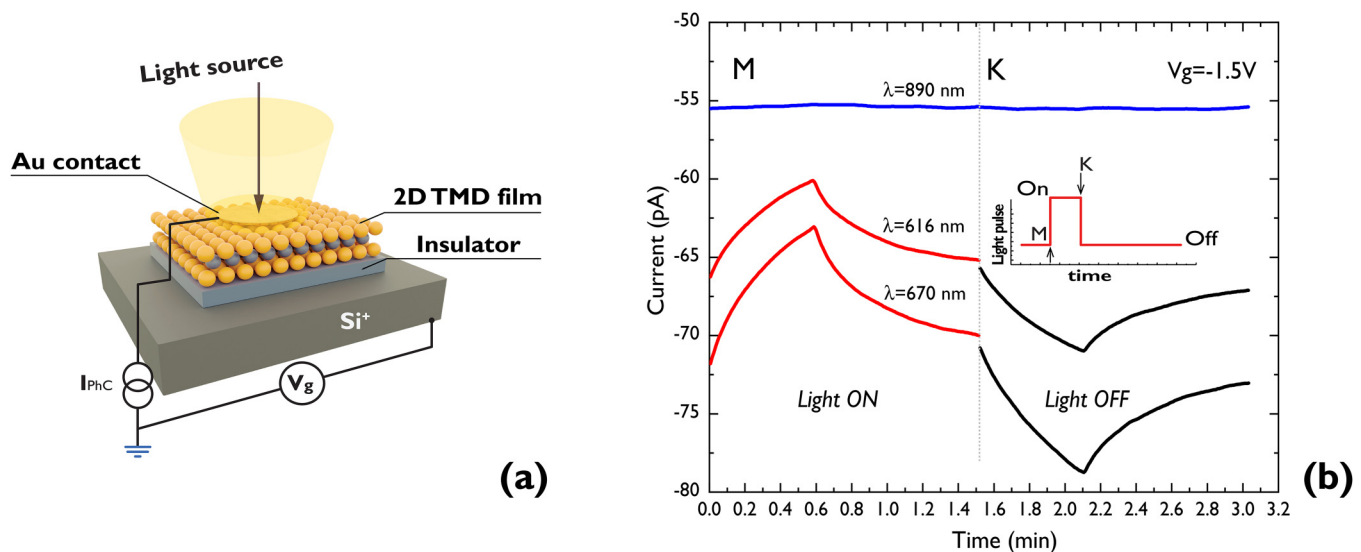


FIG. 1. (a) Schematic of a photocapacitor incorporating a transition metal dichalcogenide (TMD) transferred on top of an insulating layer. (b) The observed relaxation transient of the displacement current in an Au/3ML MoS₂/HfO₂ (12 nm)/TiN/p-type Si structure under a voltage of -1.5 V applied to the 2D layer through an Au contact, illuminated with 616, 670, or 890 nm light. The inset displays the square-shaped light pulse generated by the Xe lamp when the shutter is open. The rising and falling edges of the signal reflect the dynamics of charge trapping at the interface during transitions between illuminated and dark states. The resulting photocurrent arises from carrier transport, followed by the trapping and release (detrapping) of electrons (under negative bias) or holes (under positive bias) at the TMD/insulator interface.

by transient current behavior, observed for short time delays between opening the light shutter and reading the current [Fig. 1(b)]. Upon turning off the light, the trapped carriers are released from oxide traps at the interface, generating a current in the opposite direction and thereby inverting the overall current polarity. Our relaxation current measurements show the dependence of the signal amplitude on the excitation light wavelength [Fig. 1(b)]. Furthermore, when the illumination wavelength is downshifted to 890 nm—below the energy threshold for indirect bandgap transitions in MoS₂—the optical absorption drops dramatically. Consequently, fewer photoexcited carriers are generated, preventing the trap filling and effectively suppressing the observed photocurrent. For photon energies in the range 1.85–2 eV, close to the 3ML indirect gap bandgap, we observe an increase in transient photoconductivity. In contrast, at lower excitation energies (below 1.5 eV), the transient photoconductivity decreases. This indicates the high sensitivity of the measurements to the MoS₂ absorption spectrum.¹²

As photon energy increases, the green curve [(2) in Fig. 2(a)] reveals the IPE of electrons from the MoS₂ VB to the HfO₂ CB. Electrons absorbing photons with energy greater than 3.5 eV can overcome the MoS₂/HfO₂ barrier, contributing to the measured current. At photon energies above 5.5 eV, a further increase in the quantum yield is observed and attributed to intrinsic photoconductivity in the m-phase of HfO₂, i.e., to electron transitions from the oxide VB to the CB as schematically shown in Fig. 2(b). We elaborate this 5.5 eV threshold in Fig. 3, which is discussed below.

Intrinsic photoconductivity in TMDs requires exciton dissociation into free carriers, achieved by applying a DC electric field

across the capacitor. The photocurrent is generated by the transport and subsequent trapping or detrapping of electrons at the TMD/insulator interface ($V_g < 0$).⁴⁴ Hence, this technique allows us to identify excitonic fingerprints and study photoconductivity in TMDs.

We begin by extracting the IPE spectral thresholds from the yield spectra of 3ML MoS₂ transferred onto HfO₂, as depicted in Fig. 3(a). The $Y^{1/3}(h\nu)$ spectra exhibit a linear relationship within the energy range of 3.6–4.7 eV. This linearity enables the determination of the IPE spectral thresholds, which reflect the energy difference between the top of the MoS₂ VB and the bottom of the HfO₂ CB. The thresholds are identified by extrapolating the IPE yield to zero for each applied bias.

Initial measurements on similar capacitors indicated that the maximum safe applied bias was -2.75 V. Bias values beyond this threshold lead to the breakdown of the insulating layer. With increasing applied bias, the IPE spectra of the 3ML MoS₂ layers develop a tail that extends down to approximately 1.3 eV below the main IPE threshold. This suggests the presence of localized states in the bandgap at the interface or within the oxide layer. It has been confirmed that crystallized HfO₂ films contain deep electron-trapping gap states, which are not eliminated even after annealing at 450 °C. Photo-depopulation spectroscopy on ALD-grown HfO₂ films demonstrates that even after a 15 min anneal at 1000 °C in nitrogen, these trap states remain largely unaffected, indicating their intrinsic nature.⁵⁰

To determine the HfO₂ bandgap, we analyze the photocurrent spectral threshold using the $Y^{1/2}(h\nu)$ plot (Fig. S3 in the supplementary material), which is appropriate for indirect optical

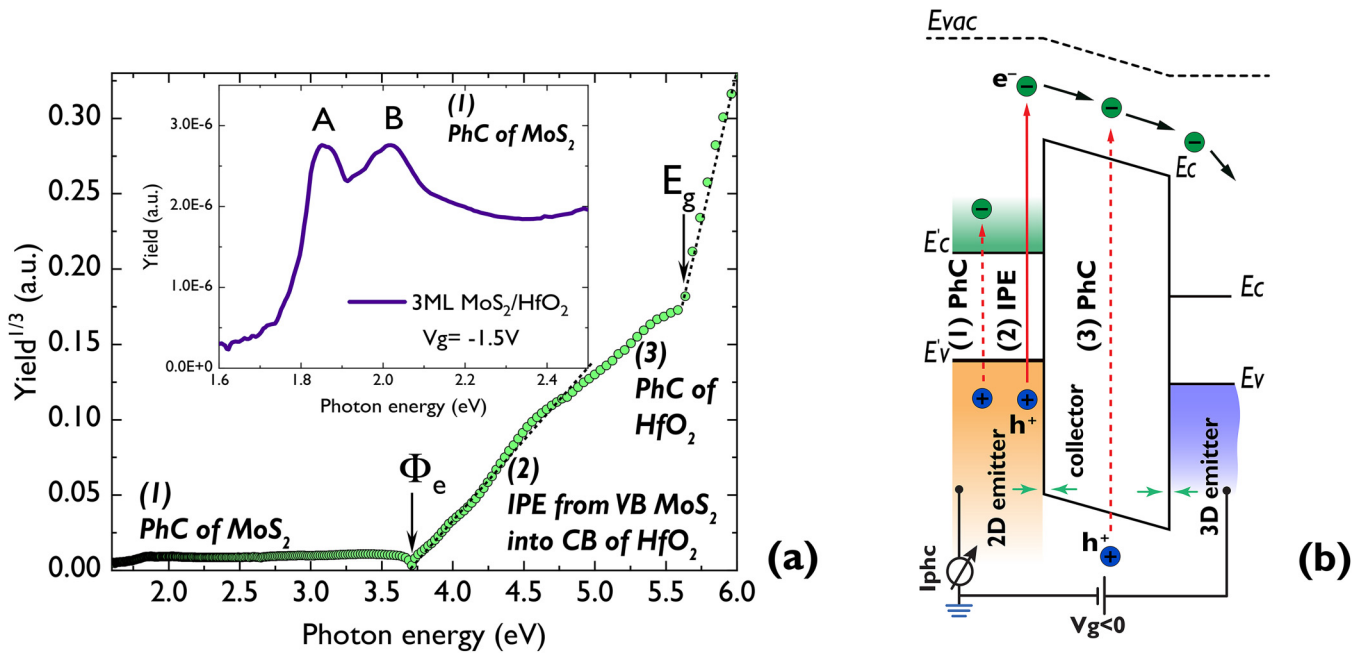


FIG. 2. (a) Cube root of photoemission quantum yield (Y) vs photon energy ($h\nu$) for the 3ML MoS₂/HfO₂/TiN/SiO₂/p-Si structure under a -1.5 V bias applied to the top Au electrode. Electrons are injected from the MoS₂ photoemitter into the HfO₂ layer. The spectrum shows several contributions: (1) photoconductivity (PhC) of 3ML MoS₂, (2) internal photoemission (IPE) from the MoS₂ valence band (VB) to the HfO₂ conduction band (CB), and (3) intrinsic photoconductivity of HfO₂. The inset highlights that in the range 1.5 – 2.5 eV, the measured quantum yield is dominated by the A and B excitons, which dissociate in the applied electric field. (b) Simplified band diagram of the 3ML MoS₂/HfO₂/TiN/SiO₂/p-Si capacitor under negative bias ($V_g < 0$). (1) The PhC contribution corresponds to electronic transitions from VB states to the CB of MoS₂. (2) The energy barrier between the MoS₂ VB and the collector CB represents the IPE threshold for electrons. (3) Oxide photoconductivity involves electron excitation to states within the collector's band structure. E_{vac} stands for electron energy in vacuum.

transitions, given that electron crystal momentum is not conserved in optical transitions in amorphous or disordered materials. This inference is consistent with the low dispersion of electron states at the top of the VB of non-magnetic oxides derived from the lone-pair O_{2p} orbitals, which makes the indirect transition model universally applicable to oxide insulators.⁴⁷ From the PhC spectra, a threshold is identified by a steep line that marks the band-to-band excitation within the 5.5 – 6 eV range in the insulator (Fig. S3 in the supplementary material). This allows us to estimate the bandgap as 5.5 eV with an experimental uncertainty of ± 0.1 eV. This gap width value is consistent with the presence of the monoclinic m-HfO₂ phase, as established previously through a combination of x-ray diffraction, optical, and photoconductivity measurements.^{51,52} The optical transitions at around 5.5 eV across the HfO₂ bandgap indicate a significant degree of crystallization in the HfO₂ layer upon annealing at 450 °C.

To determine the bandgap of the 3ML MoS₂ using the measured displacement photocurrent, we separated the excitonic and free carrier components of the photocurrent and extracted the bandgap of synthetic 3ML MoS₂ films by fitting the spectra to the Tauc model.¹² Figure S2(b) in the supplementary material shows the photocurrent spectra of a 3ML MoS₂/HfO₂/TiN/SiO₂/p-Si capacitor, measured at gate voltages of -1 and -1.5 V, normalized to the A exciton peak. An additional B peak at 2.01 eV appears

above the A peak. The Tauc plot for the 3ML MoS₂/HfO₂/TiN/SiO₂/p-Si capacitor [Fig. 3(b)] was obtained by removing the excitonic component using a linear combination of photocurrent spectra at $V_g = -1$ and -1.5 V. Since the photocurrent I_{phc} is proportional to the absorption coefficient α ,¹² the bandgap values can be extracted using the following equation:

$$(I_{phc} \cdot h\nu)^b = C \cdot (h\nu - E_g), \quad (1)$$

where $h\nu$ is the incident photon energy, E_g is the bandgap, C is a constant, and the exponent b equals 2 or $1/2$ for the direct or indirect bandgap in a 3D semiconductor, respectively. Then, E_g is extracted by extrapolating the linear region of a $(I_{phc} \times h\nu)^b$ vs $h\nu$ plot to zero current.

This procedure, validated across several capacitors with SiO₂ and HfO₂ insulators and Tauc exponents $b = 1/2$ and $b = 2$, yields indirect and direct bandgaps of 1.8 ± 0.1 and 2.4 ± 0.1 eV, respectively. From the Tauc plot and excitonic transition energies, we determine the binding energy of A exciton as $E_{bind} = E_g - E_{exc}$. For 3ML MoS₂, this gives $E_{bind} = 0.55$ eV, consistent with values of 0.3 – 0.6 eV reported for similar 3ML MoS₂ systems.^{12,53}

In multilayer MoS₂, both direct and indirect transitions arise due to its complex band structure. As the number of layers increases,

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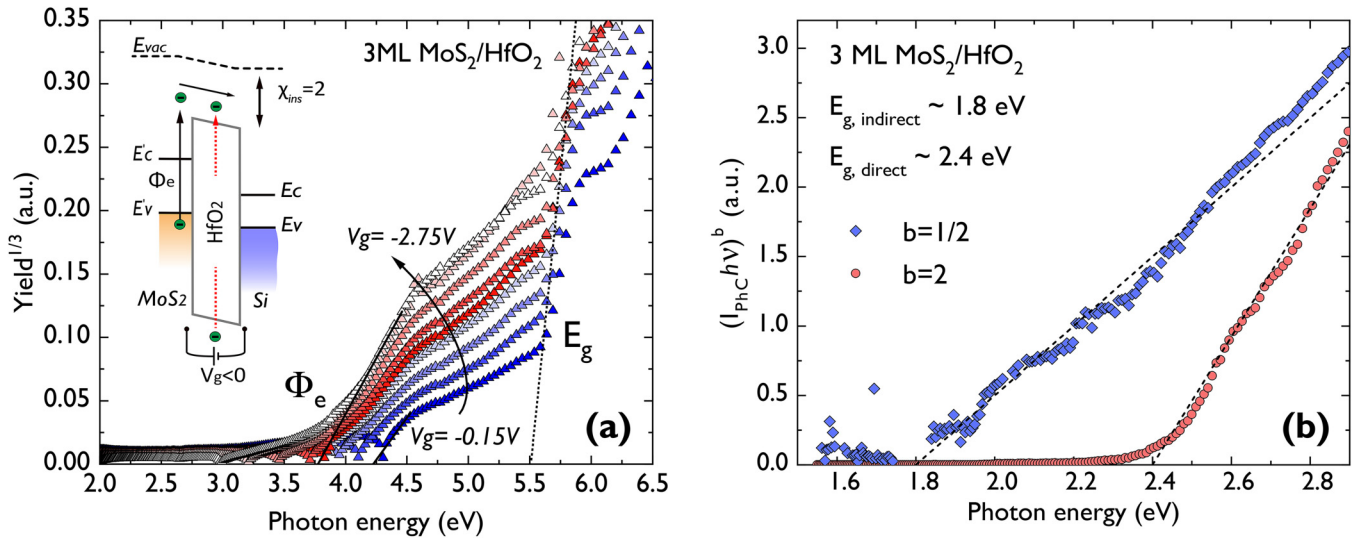


FIG. 3. (a) Cube root of the IPE yield plotted vs photon energy for 3ML MoS₂/m-HfO₂ capacitors under negative biases from -0.15 to -2.75 V. The inset shows a schematic of electronic transitions from the MoS₂ valence band (VB) to the m-HfO₂ conduction band (CB) and its photoconductivity (PhC). The vertical line marks the onset of intrinsic PhC of the oxide with $E_g(\text{HfO}_2) = 5.5 \pm 0.1$ eV. (b) Corresponding Tauc plots, where b equals 2 for direct allowed and $1/2$ for indirect allowed optical transitions. The extrapolation of the linear region for $b = 1/2$ and $b = 2$ yields 1.8 ± 0.1 and 2.4 ± 0.1 eV for the indirect and direct bandgaps, respectively.

strong spin-orbit coupling and interlayer interactions induce a transition from a direct to an indirect bandgap. Specifically, conduction band splitting along the K - Γ direction in the Brillouin zone lowers the conduction band minimum away from the K point,⁵⁴ making indirect transitions energetically favorable. Therefore, for the conduction band offset estimated from IPE, we use the indirect bandgap value of 1.8 eV, as it reflects the dominant excitation pathway and provides a realistic estimate of the carrier injection barrier at the MoS₂/insulator interface. Importantly, the energy barrier extracted from IPE measurements corresponds to the **lowest energy** required for electrons to overcome the interface barrier in the 3ML MoS₂/HfO₂ system. This barrier represents an electronic transition from the global valence band maximum at the Γ point in MoS₂ to conduction band states in the insulator. This is because photoemission is governed by the highest occupied electronic states, and in multilayer MoS₂, the absolute valence band maximum lies at Γ , not at the K point. As a result, transitions involving states at Γ dominate the onset of photoemission, even though the direct transition at K is optically allowed.

The energy of IPE thresholds Φ_e , extracted using the $Y^{1/3}(h\nu)$ plot at different applied voltages for both MoS₂/a-SiO₂ and MoS₂/HfO₂ interfaces, linearly decreases with the square root of electric field F in the oxide [Fig. 3(a)], as expected within the image-force model. This model,⁵⁵ also known as the Schottky effect, predicts that $\Phi_e = \Phi_0 - q\sqrt{\frac{qF}{4\pi\epsilon_0\epsilon_i}}$, where q is the electron charge, ϵ_0 is the vacuum dielectric permittivity, and ϵ_i is the effective image-force constant that corresponds to the dielectric function inside the barrier region and accounts for different polarization processes at the MoS₂/dielectric interface. For the relatively clean 3ML MoS₂/a-SiO₂ interface, where MoS₂ was synthesized under C-free

and H-free conditions (slow sputtering of Mo in sulfur vapor), $\epsilon_i \approx 2$ is close to $\epsilon_i \approx n^2$, where $n = 1.46$ is the refractive index of SiO₂,^{42,45,47} and the IPE barrier height is 3.6 ± 0.05 eV. Combined with experimentally determined indirect gap of 1.8 eV for 3ML MoS₂ and electron affinity of a-SiO₂ $\chi = 0.9$ eV, inferred from the combination of IPE and external photoemission spectral thresholds,⁵⁶ this results in a CB offset of ca. 1.85 eV [Fig. 4(b)]. This value closely matches the CB offset derived from the IPE threshold for the 3ML MoS₂/SiO₂ interface,^{42,45,47} considering the 1.8 eV bandgap that we determined for 3ML MoS₂ from the Tauc plots of Fig. 3(b). The χ value of 3ML MoS₂, determined using IPE barrier data, the EA of SiO₂, and the bandgap value of 3ML MoS₂, equals 2.7 eV. By contrast, much weaker barrier lowering occurs at the MoS₂/HfO₂ interface with $\epsilon_i \approx 25$. Such a weak field dependence results in an attenuated image-force interaction between a photoemitted electron and 2D photoemitter. It has been previously observed for MoS₂/a-Al₂O₃ interface⁴⁷ and indicates the presence of charge traps at the oxide surface.

The above observation can be further correlated with the corresponding sub-threshold tail in the electron IPE spectra observed for both MoS₂/HfO₂ and MoS₂/a-Al₂O₃ interfaces. As compared to the IPE barrier at the MoS₂/a-SiO₂ interface, a considerably higher IPE threshold of ≈ 4.1 eV is found for the MoS₂/HfO₂ case [Fig. 4(a)]. The $\chi_{\text{ins}}(\text{HfO}_2) = 2$ eV value can be estimated using the reported ionization energy of silicon of 5.15 eV⁵⁷ and the electron barrier at the Si/HfO₂ interface of 3.1 eV^{45,46}—the energy between the top of Si VB and the bottom edge of HfO₂ CB. Taking into account the obtained IPE threshold and the bandgap of 3ML MoS₂, the value of the CB offset at the 3ML MoS₂/HfO₂ interface can be estimated as $E_c = \Phi_e(F) - E_g \approx 2.3$ eV. When combined with the χ value of hafnia, the CB offset results in the electron affinity $\chi \approx 4.3$ eV for 3ML MoS₂.

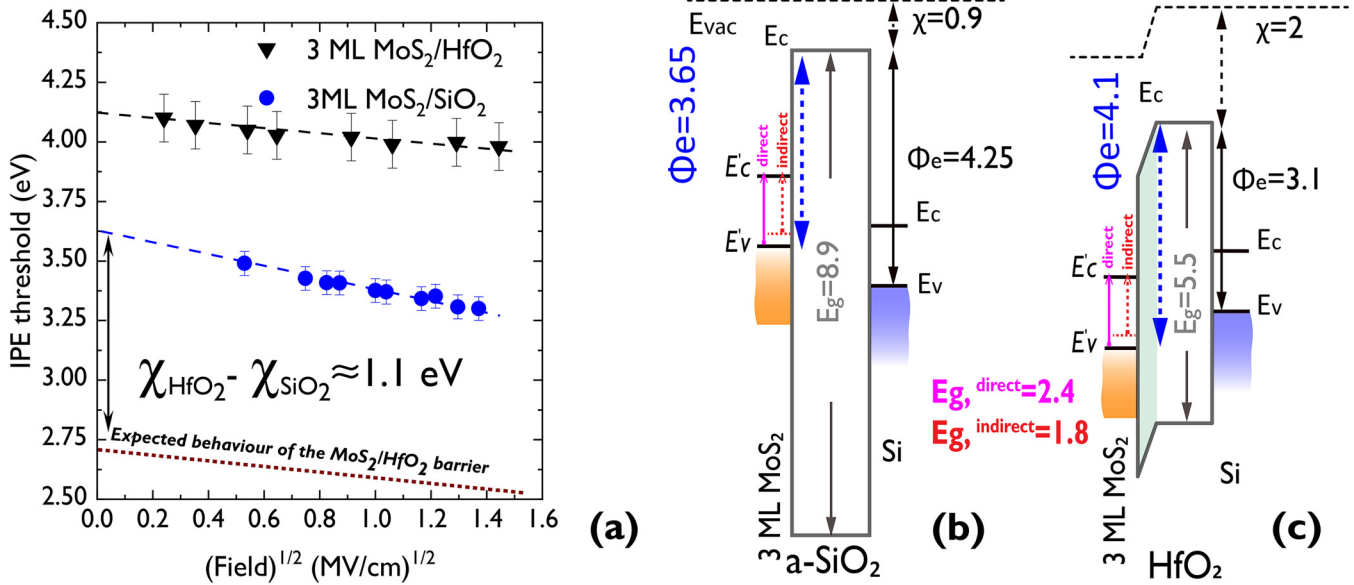


FIG. 4. (a) Schottky plots of the field dependence of the IPE thresholds. Black triangles correspond to an Au/3ML MoS₂/HfO₂ (12 nm)/TiN (20 nm)/a-SiO₂ (50 nm)/p-Si (100) capacitor, while blue circles represent 3ML MoS₂/a-SiO₂ (25 nm)/p-Si (100). Dashed lines show the linear extrapolation to zero field, and the brown dotted line marks the expected VB edge position of 3ML MoS₂ at the m-HfO₂ interface. (b) Band alignments for Si/a-SiO₂/3ML MoS₂ and (c) Si/a-SiO₂/TiN/HfO₂/3ML MoS₂ structures under flatband conditions at the Si/oxide interface, assuming no oxide charges. The measured IPE barriers (Φ_e) and corresponding electron affinities (χ) are shown for each structure. (c) The hypothesized dipole layer at the 3ML MoS₂/HfO₂ interface. All energies are in eV.

Our measurements demonstrate that, contrary to the case of chemically bonded interfaces, the difference in IPE barrier heights for 3ML MoS₂ does not correspond to the difference in the χ values of the insulating films.^{45,47,51,56} Note that the difference in

the CB offsets, given by $\Delta E_c(\text{Si/a-SiO}_2) - \Delta E_c(\text{Si/HfO}_2) = 1.15$ eV, directly reflects the variation in χ values of the two oxides predicted by the Electron Affinity Rule (EAR). This observation indicates a barrier increase of about 0.6–0.7 eV for the 3ML MoS₂/HfO₂

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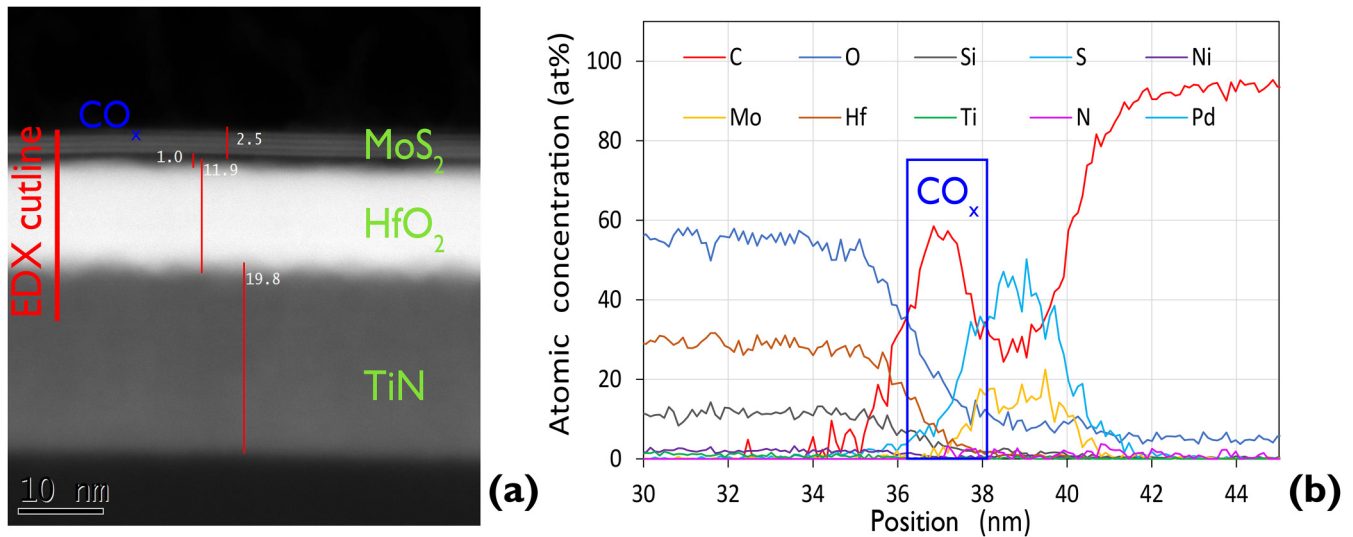


FIG. 5. (a) Cross-sectional TEM micrograph of the TiN/HfO₂/MoS₂ stack showing three complete MLs of MoS₂ with a total thickness of 2.5 nm. A carbon interlayer is seen as a dark contrast at the HfO₂/MoS₂ interface. (b) EDX analysis along the cutline confirms the presence of a 1–2 nm thick carbon-based interlayer.

TABLE I. Summary of the experimental IPE data obtained for the vdW-bonded MoS₂ interfaces as compared to the chemically bonded interfaces with silicon. The Φ_e values correspond to the electron barrier heights between the VB top in the semiconductor and CB bottom in the oxide. IPE experiments revealed no measurable dipole contribution for the MoS₂ films grown in C-free and H-free conditions using magnetron sputtering, as indicated by the “...” symbol. This observation also applies to chemically bonded interfaces. However, surface dipoles with a magnitude of 0.6–1 eV are typically observed for both sulfurization-grown and transferred MoS₂ films. PW stands for the present work.

Interface	Φ_e (eV)	Oxide E_g (eV)	Oxide χ (eV)	$q \cdot \Delta$ Vdipole (eV)	Reference
(1) Van der Waals bonded					
1ML MoS ₂ /a-SiO ₂	4	8.9	0.9	...	45
3ML MoS ₂ /a-SiO ₂	3.6	8.9		...	42
1ML MoS ₂ /a-Al ₂ O ₃	4.1	6.2	1.9	~1	45
3ML MoS ₂ /m-HfO ₂	4.1	5.5	2	~0.6	PW
(2) Chemically bonded (bulk)					
Si(100)/a-SiO ₂	4.25			...	45
Si(100)/a-Al ₂ O ₃	3.25			...	45
Si(100)/m-HfO ₂	3.1			...	46 and 47

heterojunction compared to the EAR prediction. Hence, an additional electrostatic potential variation may be present at the MoS₂/m-HfO₂ interface, which we attribute to an interface dipole layer. Therefore, the EAR needs to be corrected by including the additional term $q \Delta$ Vdipole,

$$\Delta Ec = \chi_1 - \chi_2 + q \cdot \Delta \text{Vdipole}, \quad (2)$$

where χ_1 and χ_2 are the χ values of 3ML MoS₂ and HfO₂ layers, and the term $q \Delta$ Vdipole corresponds to an additional drop of approximately 0.7 V in the electrostatic potential across the studied 3ML MoS₂/HfO₂ interface. A possible origin of this dipole is discussed below. It is important to note that a direct comparison of IPE barriers for the two cases—the direct synthesis of 3ML MoS₂ on SiO₂ and the transfer of 3ML MoS₂ onto the HfO₂ (12 nm)/TiN (20 nm)/a-SiO₂(50 nm)/p-Si (100) structure—is not straightforward. This is because, in the latter case, the energy band alignment is influenced by two factors: the intrinsic states at the interface between 3ML MoS₂ and HfO₂ (12 nm), as well as potential interface contamination resulting from the MoS₂ transfer.

To better understand the possible origin of the observed IPE barrier increase as a result of TMD layer transfer and indicate possible factors disturbing the electrostatic potential distribution across the MoS₂/HfO₂ interface, cross-sectional transmission electron microscopy (TEM) and energy-dispersive x-ray spectroscopy (EDX) analysis was performed (Fig. 5). TEM and EDX analysis of the investigated MoS₂/HfO₂/TiN/SiO₂/p-Si stack revealed the presence of a ca. 1–2 nm thick carbon-rich interlayer, probably resulting from the wet transfer processing and the use of a polymethyl methacrylate protective layer during processing. This is in stark contrast to the interface quality observed when few-ML MoS₂ was directly grown on SiO₂ by DC magnetron sputtering of Mo in sulfur vapor (Mo + S process) without the MoS₂ layer transfer (Fig. S4 in the supplementary material). Hence, the behavior of the IPE threshold at MoS₂/HfO₂ interface and the observed failure of the EAR due to the electroneutrality violation may be explained by the incorporation of carbon residuals and water interaction

with the oxide surface as a result of MoS₂ layer transfer. It suggests that the carbon residuals along with the interstitial water molecules and the surface hydroxyl (O–H) groups are the potential factors to account for the observed electroneutrality violation, which is detrimental to the electrostatics of the van der Waals bonded MoS₂/oxide interfaces.

Using the inferred IPE barriers Φ_e measured for the van der Waals bonded interfaces between 2D TMD and three different insulating oxides, and for the chemically bonded bulk interfaces, i.e., the 3ML MoS₂/a-SiO₂ vs the Si/a-SiO₂ case, the 3ML MoS₂/HfO₂ vs the Si/HfO₂ interface, etc., we constructed the band alignment schematics shown in Figs. 4(b) and 4(c). The value of the IPE barrier measured at the Si/HfO₂ interface and the Si bandgap are either taken from the literature^{45,46,55,58} or measured in the present work. As can be seen from Fig. 4, the barriers at the Si/a-SiO₂ and Si/HfO₂ interfaces are sensitive to changes in χ values of the oxide insulator, and, therefore, the EAR remains pertinent for these 3D chemically bonded heterojunctions. By contrast, the ideal affinity rule fails for both the MoS₂/a-Al₂O₃ and the MoS₂/HfO₂ interfaces by predicting significantly lower barriers than the experimentally observed ones. IPE data are summarized in Table I.

IV. CONCLUSIONS

Analysis of 3ML MoS₂ interfaces with a-SiO₂ and HfO₂ reveals a deviation from the expected ideal Schottky–Mott scenario when 3ML MoS₂ is transferred onto HfO₂. This deviation is primarily attributed to violations of interface electroneutrality, stemming from chemical modification of the oxide surface and the presence of a ca. 1 nm interlayer that contains carbon polymer and water residues from the 2D layer transfer process. Such interfacial contaminations can significantly affect the performance of TMD-based transistors, as they may lead to increased carrier scattering and the threshold voltage shift. Furthermore, the presence of these contaminants introduces electron states accessible for charge carriers in the TMD layer raising reliability concerns due to charge trapping. The IPE data indicate the existence of a polarization layer with a net negative

charge that results in an electrostatic potential variation of approximately 0.7 V across the interface. Our findings demonstrate that band alignment at 2D TMD/insulator interfaces is not only determined by the intrinsic properties of the 2D materials but is significantly affected by interfacial chemistry and processing conditions, which particularly affects the insulating oxide surface. The Schottky–Mott model and electron affinity rule are insufficient for describing these interfaces, highlighting the necessity for interface barrier height monitoring to enable interfacial engineering to control the band offsets and built-in potentials. Processing improvements, such as enhanced surface cleaning and dry transfer techniques, are essential for reducing interface contamination.

SUPPLEMENTARY MATERIAL

See the [supplementary material](#) for photoconductivity measurements and bandgap determination.

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AUTHOR DECLARATIONS

Conflict of Interest

The authors have no conflicts to disclose.

Author Contributions

I. Shlyakhov: Conceptualization (lead); Data curation (lead); Formal analysis (lead); Investigation (lead); Methodology (lead); Writing – original draft (lead); Writing – review & editing (lead). **K. Iakoubovskii:** Conceptualization (equal); Methodology (equal); Writing – review & editing (equal). **D. Lin:** Conceptualization (equal); Methodology (equal). **I. Asselberghs:** Supervision (supporting). **A. Gaur:** Investigation (supporting). **G. Delie:** Investigation (supporting). **V. Afanas'ev:** Conceptualization (equal); Project administration (lead); Supervision (lead); Writing – review & editing (equal).

DATA AVAILABILITY

The data that support the findings of this study are available from the corresponding author upon reasonable request.

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