

1 Consequences of composition on morphological and mechanical
2 properties in fully renewable poly(lactide)-poly(farnesene) block
3 copolymers

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12 **Keywords:** *Sustainable polymers; renewable thermoplastic elastomers; self-assembling block*
13 *polymers; flow polymerization; poly(farnesene); poly(lactide)*

14 **ABSTRACT**

15 Reliance on non-renewable fossil resources, sluggish degradability, and limited recycling
16 opportunities encourage the implementation of more sustainable materials and manufacturing
17 practices in society. Poly(lactide) (PLA), an FDA-approved biobased polymer with biodegradable
18 properties has long been studied as a promising alternative. However, there is a need to
19 overcome the inherent brittleness associated with PLA, thereby drastically improving its
20 applicability. We explored the incorporation of hydroxyl-telechelic poly(β -farnesene) (PF) - a
21 biobased, hydrophobic polymer derived from terpenes - as a soft midblock in a PLA-PF-PLA
22 triblock system. The effect of molecular weight and weight fraction of PLA on morphology was
23 studied using SAXS and TEM. Lower molecular weight polymers assembled into a lamellar
24 morphology at high PLA weight fractions, while high molecular weight polymers adopted
25 hexagonally-packed cylinder morphology, and exhibited relatively elastomeric behavior. Tensile
26 studies revealed that mechanical properties can be tuned by altering the PLA composition, with
27 higher weight fractions increasing tensile modulus to 22.1 MPa. Additionally, a comprehensive E-

28 factor analysis was performed on the block copolymer synthesis in order to highlight the
29 importance of process optimization when designing complex sustainable polymers.

30

31 **INTRODUCTION**

32 Plastics are found in nearly every consumer product today, from packaging materials and
33 electronic devices to clothing, due to their straightforward production, ease of processing and
34 exceptional durability.¹⁻⁴ However, nearly all plastics are derived from non-renewable fossil
35 sources, are only degradable over vast timescales, and are prohibitively difficult to recycle. We
36 are now in the midst of a plastic pollution crisis and are continuing to deplete precious resources.^{1,}
37 ^{3, 5} Making substitutions to our commodity polymer platforms with renewable (and/or recyclable)
38 alternatives remains a challenge, predominantly because property matching is difficult.

39 Exploring renewable, biobased alternatives for plastic materials remains one of the critical
40 strategies in transitioning to a circular economy.⁴⁻¹⁰ Extensive efforts have been made to introduce
41 poly(lactide) (PLA), a bio-based polymer, into everyday plastic products.¹¹⁻¹⁴ PLA is typically made
42 by ring-opening transesterification polymerization (ROTEP) of lactide, which itself is obtained
43 through the anaerobic fermentation of carbohydrates.^{15, 16} Additionally, the FDA has approved
44 PLA for food packaging and medical materials such as sutures.¹⁶ Lastly, PLA is a polyester, and
45 can therefore biodegrade under composting conditions without any residual microplastics,
46 tackling a growing concern in the field of environmental health sciences.¹⁶⁻¹⁸

47 However, PLA homopolymer is notoriously brittle.^{15, 19} Expanding the range of applications
48 for which PLA is suitable can be achieved with block polymers, where the relatively rigid PLA
49 typically acts as a hard block.²⁰⁻²⁶ Combining PLA with a softer segment (i.e., poly(butadiene),
50 poly(isoprene)), typically leads to improved elasticity and impact resistance.²⁷⁻³⁰ This in turn
51 modifies the overall polymer behavior to exhibit a wide range of properties depending on the
52 respective block compositions.³¹⁻³⁵ Fully renewable PLA-based block polymers with properties
53 targeting elastomeric behavior have been reported, typically utilizing structurally related aliphatic

54 polyester soft-blocks.³⁶⁻⁴⁰ However, the number of suitable biobased soft-blocks that are
55 amenable to chain extension with PLA remains relatively limited.⁴¹

56 One potentially appealing candidate is poly(β -farnesene), a biobased, hydrophobic
57 polymer derived from terpenes, which in turn are a class of natural molecular biomass produced
58 by various plants and animals. They are also a major component of tree resin, an exudate typically
59 obtained from pine trees and conifers.⁴² Terpene derivatives such as isoprene, myrcene and
60 farnesene have garnered attention in the field of elastomers thanks to their relatively facile
61 polymerization and tuneability, in addition to being renewable alternatives to commercial
62 petrochemical based diene counterparts.⁴³⁻⁴⁶

63 In this study, PLA was incorporated into symmetric ABA triblock copolymers by applying
64 hydroxyl-telechelic poly(β -farnesene) as a macroinitiator for the ROTEP of lactide with the aim of
65 accessing elastomeric materials. To investigate the relationship between the overall molar mass
66 and respective PLA weight compositions, morphological-, thermal-, and mechanical properties
67 were systematically evaluated for the first time. Establishing the connection between mechanical
68 properties and molecular attributes in these fully renewable copolymers highlights some of the
69 opportunities for various applications and challenges that remain for expanded utility.

70

71 **RESULTS AND DISCUSSION**

72 This work leverages the insights from our previous publication, in which poly(farnesene)- and
73 poly(lactide)-based triblock copolymers were prepared using an unconventional plug flow
74 reactor.⁴⁴ We previously showed an efficient synthetic route to prepare differently composed
75 triblocks to augment their mechanical attributes for a broader spectrum of applications.
76 Previously, we reported the synthesis of symmetric ABA-type triblock copolymers using two
77 different samples of poly(farnesene)-diol macroinitiators (4 kg/mol and 30 kg/mol). However, as
78 the focus was primarily on the implementation of various reaction parameters for block copolymer

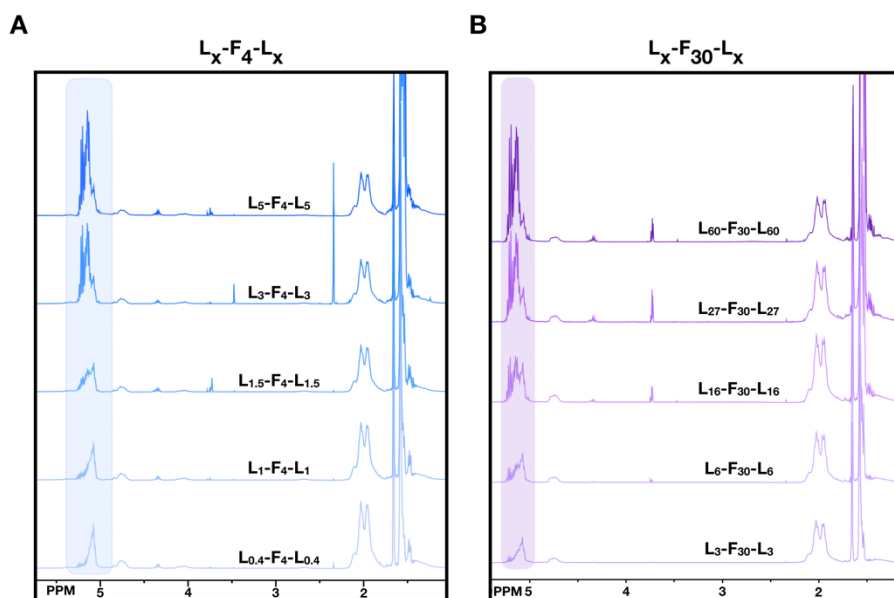
79 synthesis using a novel flow reactor, a thorough morphological- and mechanical analysis was not
80 in scope. In this study we prepared a new series of copolymers with various chain-lengths and
81 compositions (i.e., A:B ratios). A series of samples was prepared covering a composition range
82 for poly(lactide) of 20 wt %, 30 wt %, 50 wt %, 70 wt %, and 80 wt %, using the two
83 poly(farnesene)-diol macroinitiators. The resulting block copolymers are hereafter referred to as
84 $L_x-F_y-L_x$, where x and y reflect the molar mass of PLA and poly(farnesene) (in kg/mol),
85 respectively. Subsequently, the molecular and physical properties of these samples were
86 interrogated with respect to their composition and molar mass. Solution polymerization was
87 selected over bulk polymerization to ensure sufficiently low viscosity for continuous flow operation.
88 While the planar flow reactor is capable of handling viscous systems, the introduction of solid
89 reagents such as lactide presents practical challenges. In terms of sustainability, flow
90 polymerizations offer some appealing features, assuming that the solvent employed can be
91 efficiently recovered (vide infra, E-factor analysis). In batch, solution polymerization additionally
92 provided enhanced mixing, thereby allowing more precise control over molecular weight and block
93 architecture. For the monomer choice, DL-lactide was employed instead of L-lactide.
94 Polymerization of DL-lactide yields amorphous PLA segments, which not only facilitates
95 processing (e.g., melt-pressing at lower temperatures) compared to isomerically pure systems
96 prone to crystallization, but also allows the morphological features to be assessed that result from
97 the mixing thermodynamics rather than induced by crystallization.

98

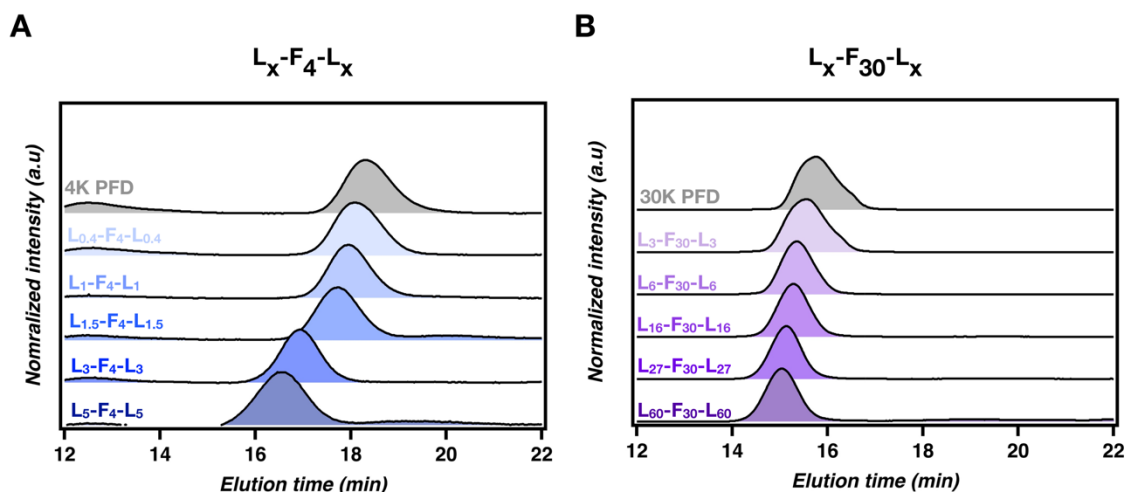
99 **PLA composition and molar mass distribution:**

100 Compositions of the block polymers were determined with ^1H nuclear magnetic resonance (NMR)
101 spectroscopy (Figure 1). The relative content of PLA repeating units was calculated as the ratio
102 of methine protons in PLA to aliphatic repeating units in the poly(farnesene) side branches (Figure
103 S2). Additionally, size exclusion chromatography (SEC) was used to determine the molar mass,
104 dispersity, and composition of the polymers (Figure 2; Table 1). A quadruplet signal is present at

105 δ 4.4 ppm for every sample, with the intensity increasing for higher PLA compositions. This peak
106 corresponds to the PLA end groups and increases with intensity as the number of PLA chains
107 increases, presumably arising primarily from transesterification. Increased transesterification is
108 consistent with the increase in dispersity for higher PLA compositions in high molecular weight
109 samples. Additionally, a distinct peak between δ 3.6 – 3.8 ppm can be seen in the higher
110 molecular weight samples as PLA composition increases. This peak is assigned to methyl-ester
111 chain ends that arise from limited methanolysis that occurs during precipitation.
112



113
114 **Figure 1.** ¹H NMR spectra of (A) the shorter and (B) the longer analogs of LFL block copolymers.
115



116
117 **Figure 2.** SEC chromatograms for (A) lower and (B) higher molar mass LFL block copolymer
118 series.

119
120 **Table 1.** Summary of ^1H NMR and SEC analysis on low- and high molecular weight block
121 copolymer samples.

	Sample name	Target ϕ_L	M_w (kg/mol) (SEC) ^a	M_n (kg/mol) (SEC) ^a	ϕ_L (^1H NMR) ^b	ϕ_L (SEC) ^c	\bar{D}^a
PFD (4 Kg/mol)	L _{0.4} -F ₄ -L _{0.4}	0.2	5.1	4.9	0.22	0.16	1.17
	L ₁ -F ₄ -L ₁	0.3	6.6	6.3	0.31	0.35	1.17
	L _{1.5} -F ₄ -L _{1.5}	0.5	7.2	7.0	0.52	0.41	1.30
	L ₃ -F ₄ -L ₃	0.7	10.2	10.1	0.64	0.58	1.16
	L ₅ -F ₄ -L ₅	0.8	13.6	13.6	0.76	0.69	1.17
PFD (30 Kg/mol)	L ₃ -F ₃₀ -L ₃	0.2	43.1	43.1	0.19	0.21	1.15
	L ₆ -F ₃₀ -L ₆	0.3	47.7	47.7	0.34	0.28	1.39
	L ₁₆ -F ₃₀ -L ₁₆	0.5	67.5	67.4	0.52	0.49	2.28
	L ₂₇ -F ₃₀ -L ₂₇	0.7	90.4	89.4	0.73	0.62	3.39
	L ₆₀ -F ₃₀ -L ₆₀	0.8	158.2	158	0.81	0.78	1.27

122 ^a Determined from SEC in chloroform from multi-angle laser light scattering detector, with dn/dc
123 calculated internally based on sample concentration and assuming 100% mass elution. ^b wt %
124 composition of L-block determined from relative integration of ^1H NMR spectroscopy ^c wt %
125 composition as determined indirectly from the molar mass from SEC.
126

127 Experimentally determined compositions closely matched the theoretical targets for both the high
128 molar mass and low molar mass sample sets (Table 1). The minor deviations in PLA compositions
129 between SEC and ^1H NMR analyses can be attributed to lower molecular weight polymer chains

130 resulting from transesterification. These small signals presumed to be from homopolymer were
131 not integrated during the SEC analysis, yet still contribute to ^1H NMR signals. We therefore expect
132 the PLA contribution in ^1H NMR analysis to be slightly higher, especially in higher PLA weight
133 contribution samples.

134 The low dispersities observed for all samples are indicative of a controlled polymerization
135 process. Furthermore, SEC analyses indicated a subtle deviation from the target PLA composition
136 at higher PLA contributions (this is especially evident in L₃-F₄-L₃, L₅-F₄-L₅ and L₂₇-F₃₀-L₂₇). Higher
137 PLA content corresponds with higher molecular weight, which can influence the conversion rate.
138 Specifically, the growth in chain lengths of the polymer can impede diffusion toward reactive chain
139 end-groups and reduce the mobility of lactide monomers which also decrease in concentration,
140 slowing the reaction kinetics. In addition to the factors mentioned, it's important to consider that
141 ROTEP is typically a reversible (i.e., equilibrium) process. Higher molecular weight polymers tend
142 to shift the equilibrium toward polymerization products, thus potentially decelerating the overall
143 reaction rate and result in smaller block copolymer fragments that are omitted in the SEC analysis.

144 ^{47, 48}

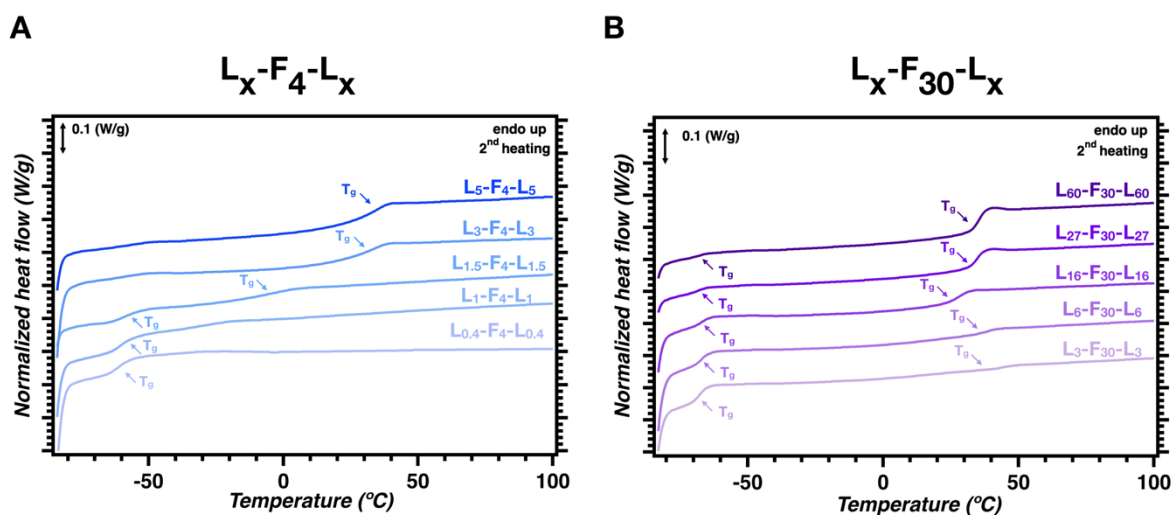
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146 **Analysis of thermal properties:**

147 All polymer samples were analyzed via differential scanning calorimetry (DSC) to investigate their
148 thermal properties and correlate them to molecular characteristics. The glass transition (T_g) of
149 PLA consistently increases with increasing molar mass in the shorter homologues, consistent with
150 expectations (Figure 3a; Figure S4).⁴⁹ The distinct T_g observed for the individual blocks suggests
151 a relatively strong immiscibility between the two blocks, which is consistent with previous reports
152 on PLA block polymers having a relatively hydrophobic, hydrocarbon based block.^{29, 50, 51} The
153 notable exception to this is sample L_{0.4}-F₄-L_{0.4}, which is not entirely surprising based on the very
154 short PLA segments. This shortest block polymer has only 3 repeating units of PLA per segment

155 on average. Such short blocks are likely nearly miscible with the hydrophobic PFD blocks, which
156 is consistent with the X-ray analysis (*vide infra*).

157



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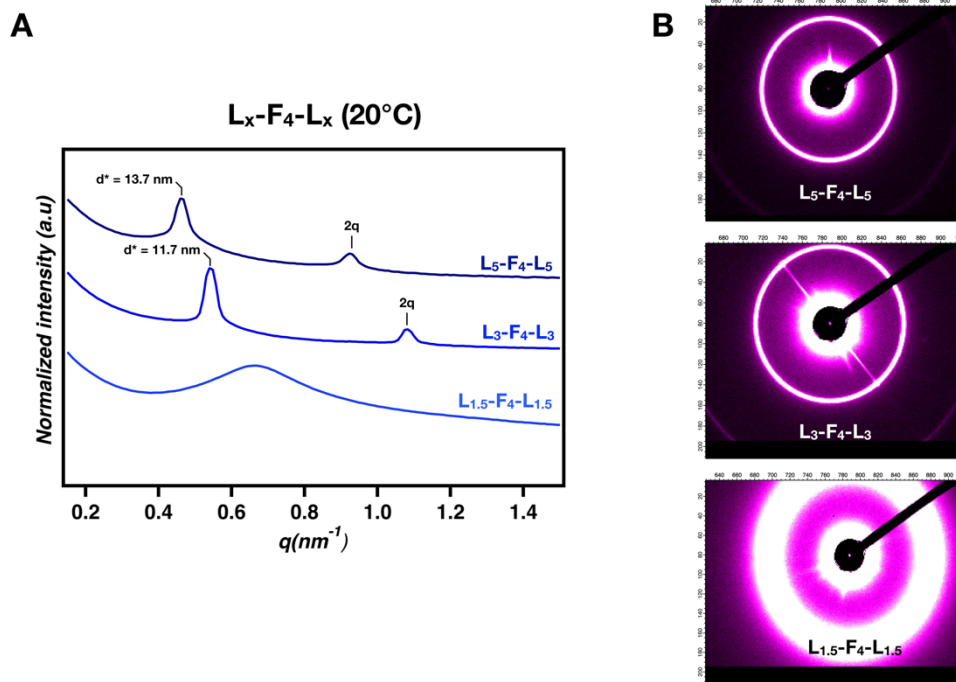
159 **Figure 3.** DSC thermograms of LFL block copolymers for (A) lower and (B) higher molar mass
160 homologues.

161

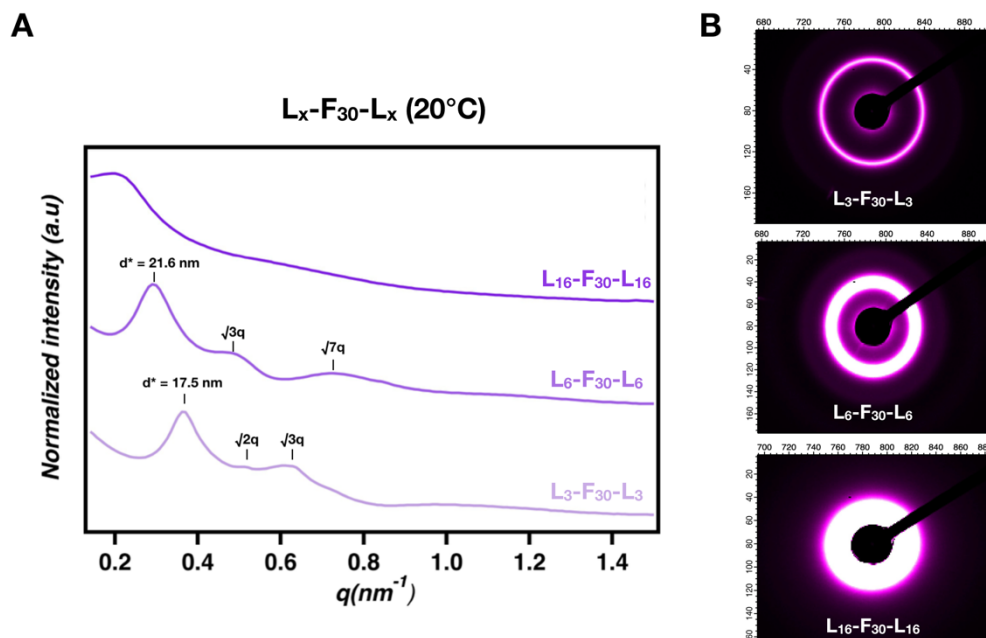
162 Morphological behavior

163 The meso-scale morphological characteristics and domain spacing were determined at various
164 temperatures (20–140 °C) using small-angle X-ray scattering (SAXS). The results indicate a clear
165 interplay between PLA composition, poly(farnesene) molecular weight, and temperature in
166 relation to microphase organization. In the case of the shorter chain sample set ($L_x-F_4-L_x$), phase-
167 separated structures were observed, albeit lacking long-range order. This was the case for
168 compositions of ϕ_L lower than 70 wt % due to the small number of PLA repeating units per
169 segment. On the other hand, higher PLA compositions ($L_3-F_4-L_3$ and $L_5-F_4-L_5$) exhibit patterns
170 consistent with lamellar organization, as evidenced by sharp signals at q^* and $2q$. The position of
171 the principle scattering vectors q^* correspond with domain spacings (d^*) of 11.7 nm and 13.7 nm
172 for $L_3-F_4-L_4$ and $L_5-F_4-L_5$, respectively (Figure 4 A,B). These peaks correspond to the 100 and 200
173 planes of alternating lamellae.⁵²⁻⁵⁴ Conversely, in the case of higher molecular weight samples L_x-

174 F₃₀-L_x, the samples L₃-F₃₀-L₃ and L₆-F₃₀-L₆ exhibit pronounced peaks at q^* and $\sqrt{3}q$,
 175 corresponding to principal domain spacings of 17.5 nm and 21.6 nm, respectively (Figure 5 A,B).
 176 These patterns are characteristic of hexagonally packed PLA cylinders in a poly(farnesene)
 177 matrix.^{52, 55}
 178



179
 180 **Figure 4:** SAXS summary of low molecular weight LFL block copolymers at 20°C (A), including
 181 the corresponding 2D patterns used for azimuthal integration (B).
 182



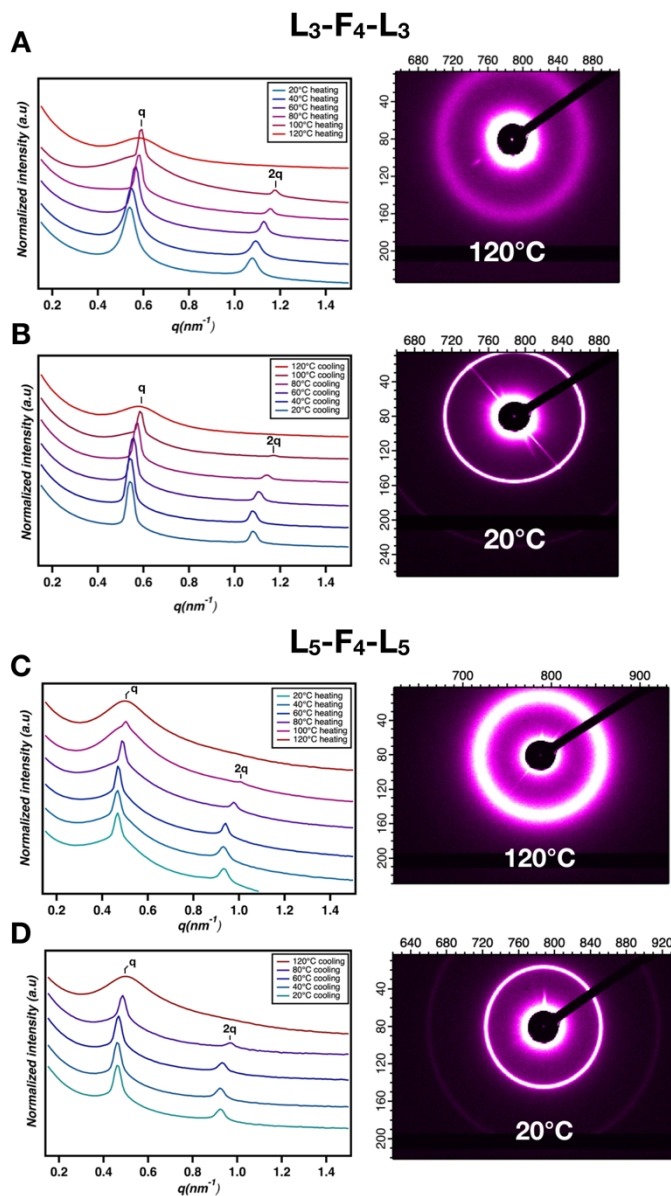
183
 184 **Figure 5:** SAXS summary of high molecular weight LFL block copolymers at 20 °C (A), including
 185 the corresponding 2D patterns used for azimuthal integration (B).
 186

187 The self-assembly of block copolymers is governed by thermodynamic considerations, including
 188 the Flory-Huggins interaction parameter (χ), which quantifies in an empirical manner the free
 189 energy cost per monomer in a situation when two unlike units want to segregate and can be
 190 estimated by models such as Random Phase Approximation (RPA) or self-consistent field theory
 191 (SCFT).^{31, 56, 57} The immiscibility of the different polymer segments is the essence of the driving
 192 forces for microphase segregation.⁵⁸ The combination of relative composition and molar mass
 193 essentially determines the morphology that emerges, such as lamellar and [hexagonally packed]
 194 cylindrical structures, as well as the size of the domains and their spacing.⁵⁸ For low molecular
 195 weight LFL, microphase separation occurs only at high PLA compositions (> 50 wt %) as lower
 196 compositions have only a few repeating units of PLA on each side of the PF block, and thus mixing
 197 (i.e., disorder) is favored.⁵⁹ Higher PLA compositions (L₃-F₄-L₃ and L₅-F₄-L₅), however, show
 198 scattering peaks (q , $2q$) that are indicative of a lamellar morphology. The relatively flat interfaces
 199 associated with lamellae are adopted despite the strong asymmetry in the composition of these

200 samples. This is most likely related to two mutually reinforcing effects: the conformational
201 asymmetry typical of PLA-containing blocks^{60, 61}, and the short-chain branching of the PFD
202 midblock.⁶² The observation of compositionally asymmetric lamellae in block copolymers with
203 branched architectures in one of the blocks is in line with both SCFT models and experimental
204 results. Briefly, the boundaries between morphologies in the phase diagram are predicted to shift
205 substantially compared with classical diblock copolymers.^{63, 64} For example, lamellae are
206 predicted to form in a block polymer that is rich in the linear component (i.e., PLA). Highly
207 asymmetric lamellae were observed experimentally in branched PLA-based block polymers in the
208 past.^{65, 66} In contrast, volumetric asymmetry in high molecular weight LFL samples results in
209 microphase separation in lower PLA weight fractions (i.e., $\phi_L < 50$). Larger PF blocks relax by
210 curving the interface towards shorter PLA blocks, forming hexagonally packed cylinders of PLA
211 in a matrix of PF.^{60, 67-71}

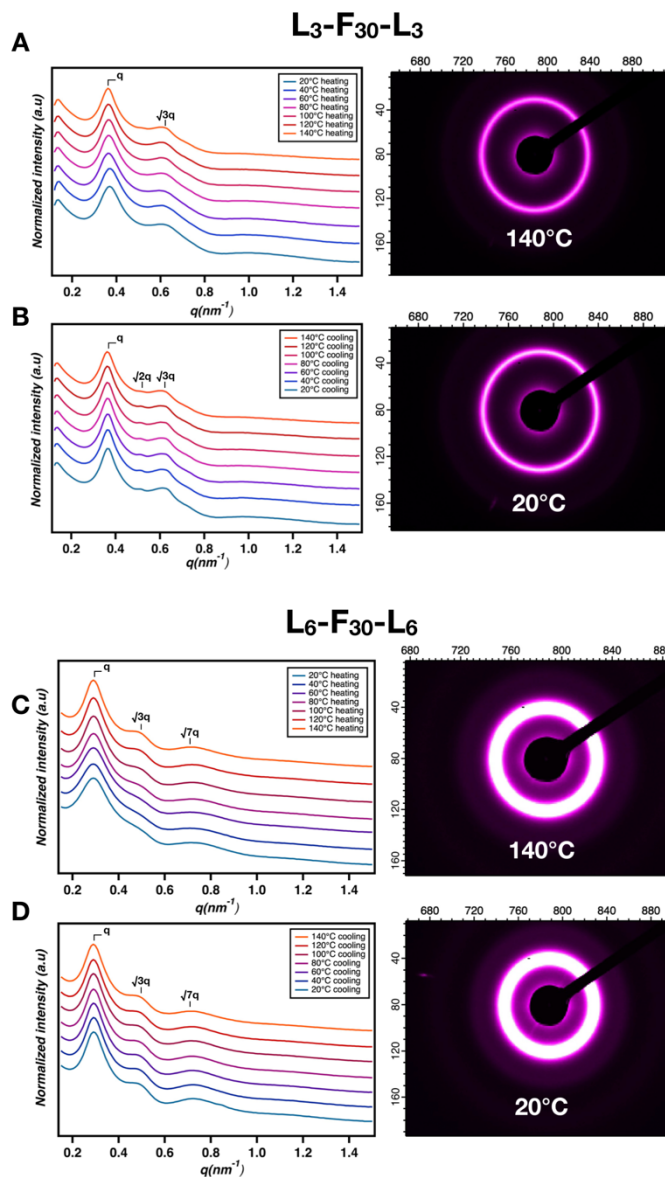
212 Individual LFL block copolymer samples were also analyzed by SAXS at various
213 temperatures, heating the sample from 20 °C to 120 °C. For lower molecular weight polymers that
214 exhibit microphase separation (L₃-F₄-L₃ and L₅-F₄-L₅), significant broadening of the peaks
215 occurred at 80 °C for both samples, attributed to increased interphase diffusion, and less definition
216 in the interface between the microdomains. At 120 °C, neither sample displays long range
217 organization. This can also be seen on the 2D SAXS images where the concentric rings become
218 broadened at higher temperatures (Figure 6 A,C). Upon cooling, both samples regain the
219 organized morphologies. Additionally, the increased rigidity of the polymer chains fixes the
220 microphase conformations as they become more constrained. As a result, the peaks
221 corresponding to the lamellar micro-domains become narrower and the 2D SAXS image becomes
222 slightly more intense (Figure 6 B,D). The effect of heating and consequent cooling is the same for
223 higher molecular weight samples with low PLA compositions (L₃-F₃₀-L₃ and L₆-F₃₀-L₆), although
224 the peak broadening is less pronounced, even at 140 °C, suggesting that the increased molecular
225 weight of the sample significantly retards molecular diffusion (Figure 7). Remarkably, in the high

226 molecular weight L₃-F₃₀-L₃ sample, a small but prominent peak at $\sqrt{2}q$ is consistent with body-
 227 centered cubic spheres (BCC spheres) with a domain spacing of 12.5 nm.⁷² This is most likely
 228 due to annealing, whereby extended periods at high temperatures can promote metastable
 229 phases that do not appear when heating the sample.^{73, 74}
 230



231
 232 **Figure 6:** Summary of simultaneous heating- / cooling cycles and SAXS analysis for low
 233 molecular weight LFL block copolymers. (A,B) 1D and 2D SAXS graph for heating and cooling of

234 L₃-F₄-L₃, respectively. (C,D) 1D and 2D SAXS graph for heating and cooling of L₅-F₄-L₅,
 235 respectively.
 236

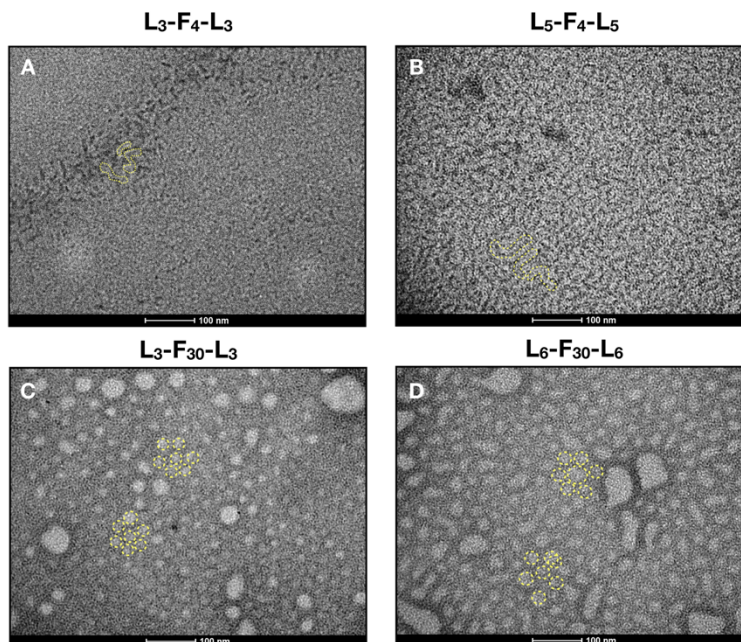


237
 238 **Figure 7:** Summary of simultaneous heating- / cooling cycles and SAXS analysis for high
 239 molecular weight LFL block copolymers. (A,B) 1D and 2D SAXS graph for heating and cooling of
 240 L₃-F₃₀-L₃, respectively. (C,D) 1D and 2D SAXS graph for heating and cooling of L₆-F₃₀-L₆,
 241 respectively.
 242

243 **Transmission electron microscopy (TEM):**

244 All samples that showed microphase separation based on the SAXS analysis were visually
245 analyzed using transmission electron microscopy (TEM). For the low molecular weight sample
246 series, worm-like structures were observed for L₃-F₄-L₃ and L₅-F₄-L₅, consistent with the lamellar
247 microphase organization displayed in the SAXS profiles (Figure 8 A,B). For L₅-F₄-L₅, it was clear
248 that the lamellar organization was more pronounced, with the average lamellar thickness
249 increasing from 6.2 nm for L₃-F₄-L₃ to 10.6 nm for L₅-F₄-L₅. The increase in lamellar thickness
250 agrees with the SAXS data. However, the domain sizes differ slightly yet fall within the same
251 range. Additionally, the TEM analysis suggests the absence of long-range order, as the lamellar
252 worm-like structures are randomly oriented in the L₅-F₄-L₅ samples and very localized in L₃-F₄-L₃.
253 Nevertheless, average dimensions measured from micrographs are consistent with the SAXS
254 measurements and corroborate the morphology. Likewise, samples L₃-F₃₀-L₃ and L₆-F₃₀-L₆
255 showed microphase separation that supported the respective SAXS data (Figure 8 C,D).
256 Localized hexagonally packed circular patterns were present in both samples which were lighter
257 in color than the surrounding matrix, indicating that the circles were composed out of PLA. For L₃-
258 F₃₀-L₃, the average center-to-center distance between cylinders was 24.5 nm with an average
259 cylinder diameter of 11.7 nm. These dimensions increased in size with increasing PLA content
260 (L₆-F₃₀-L₆), with the average center-to-center distance between cylinders being 34.5 nm with an
261 average diameter of 22.3 nm. Although the images suggest that the block copolymer is
262 microphase separated, the lack of long-range order suggested by SAXS profiles is corroborated
263 by the TEM images.

264



265
 266 **Figure 8:** TEM analysis of LFL block copolymers at 135Kx zoom. (A,B) low molecular weight LFL
 267 displaying localized lamellar microphase separation. (C,D) high molecular weight LFL displaying
 268 hexagonally packed cylinders.

269
 270 **Table 2:** Summary of dimensional analysis performed on the TEM images of different LFL block
 271 copolymer samples.

PFD molar mass (kg/mol)	PLA composition (wt %)	Inter-cylinder space (nm)	Cylinder diameter (nm)	Lamellar thickness (nm)
4	70	/	/	6.2 ± 1.1
4	80	/	/	10.6 ± 2.0
30	20	24.7 ± 2.6	11.7 ± 1.7	/
30	30	34.5 ± 4.4	22.3 ± 2.6	/

272
 273 Notably, mesoscale phase separation was also observed in higher PLA compositions for high
 274 molecular weight poly(farnesene), albeit lacking any long-range order. It is difficult to conclude
 275 any organized morphology, probably owing to the higher level of entanglements, large
 276 incompatibility between PLA and PFD blocks, and lack of annealing employed (Figure S5).

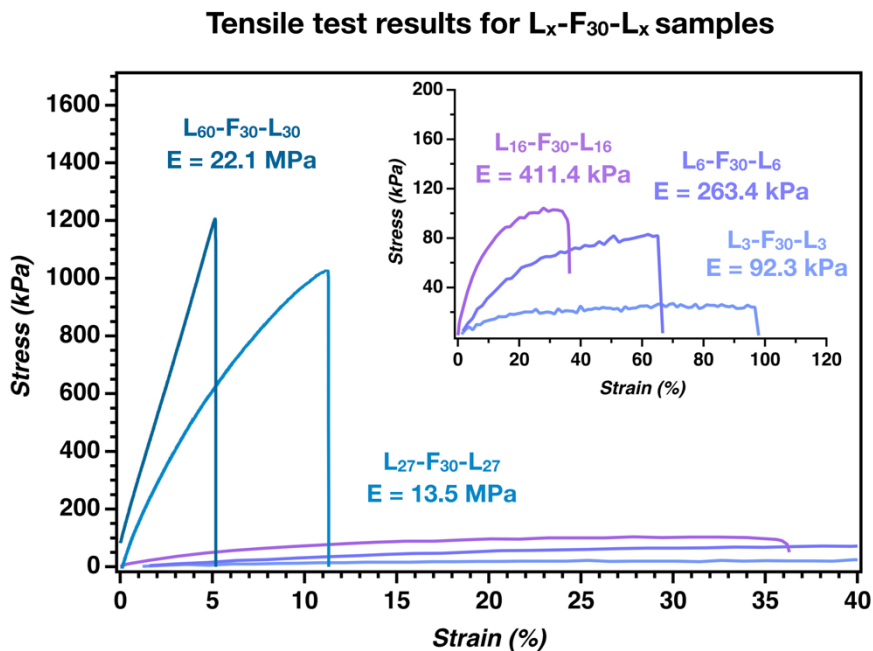
277 Nevertheless, the dimensions are significantly larger, commensurate with the higher molar
278 masses.

279

280 **Mechanical analysis of high molecular weight LFL block polymers**

281 Selected $L_x-F_{30}-L_x$ samples were subjected to tensile testing experiments, whereby the melt
282 compressed molded samples were cut into dog-bone shapes (35 mm × 2 mm × 1 mm) and
283 mechanically extended until they broke. $L_3-F_{30}-L_3$ exhibited a low modulus (92.3 kPa) and broke
284 at relatively high strain (98%). $L_6-F_{30}-L_6$ has a higher modulus in addition to a reduced strain.
285 Lastly, when the PLA content is increased to 50 wt %, stiffness is further increased, and the strain
286 at break is reduced to 38%. The increasing brittleness is coupled with a drastic decrease in
287 flexibility as the Young's modulus increases to 411.4 kPa (Figure 9). Samples with very high PLA
288 content, $L_{27}-F_{30}-L_{27}$ and $L_{60}-F_{30}-L_{60}$, far exceeded the Young's modulus of lower composition
289 samples (13.5 MPa and 22.1 MPa, respectively) but compromise in their elastomeric behavior
290 owing to the low strain at break. As the poly(farnesene) component diminishes, the overall
291 mechanical properties will more closely resemble those of high molecular weight PLA
292 homopolymer (i.e., a very tough, but brittle material).^{25, 27, 30} The very low volume fractions of soft-
293 block material in $L_{27}-F_{30}-L_{27}$ and $L_{60}-F_{30}-L_{60}$ allow for very subtle flexibility before breaking at
294 minimal deformation. Referring to literature, numerous PLA based block copolymers designs have
295 been investigated for their mechanical properties. Comparing the data for a selection of materials
296 highlighted in a recent perspective study, it becomes clear that our PLA-based copolymer design
297 showcases modest mechanical properties.²⁵ We postulate this behavior to originate from several
298 molecular features. Firstly, as mentioned, we used racemic DL-lactide instead of isomerically pure
299 L-lactide as a monomer and assumed that the amorphous nature of the resulting PLA chain would
300 enhance elasticity without compromising toughness. However, the lack of crystallinity leads
301 generally to weak and brittle materials. The lack of mechanical integrity is likely exacerbated, as

302 steric repulsion from sidechains on the PF midblock may inhibit chain entanglement and therefore
303 reduce resistance to fracture.



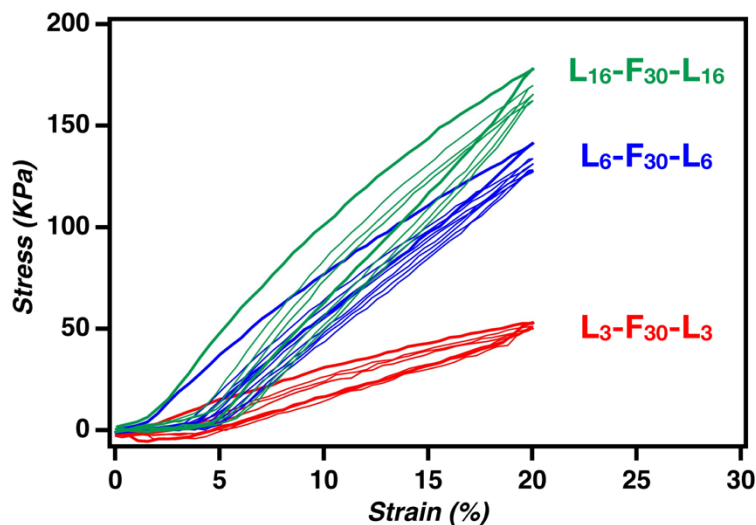
304
305 **Figure 9:** Stress/strain curves for high molecular weight LFL block copolymers

306
307 Additionally, cyclic tensile tests were performed on the L_x-F₃₀-L_x polymers with 20%, 30%, and
308 50% PLA compositions in order to test the durability of the mechanical properties. All samples
309 were elongated to 20% strain at 100 mm per minute. The recorded stress in kPa after each cycle
310 provided insight into the decrease of the Young's modulus for each sample. Lower PLA
311 compositions (L₃-F₃₀-L₃) exhibit very elastic behavior owing to the higher contribution of the soft
312 block component, with the hard PLA segment providing minimal integrity, yet enough to allow for
313 elastic characteristics, as well as a minimal loss of mechanical properties over repeated
314 stretching. This contrasts with increased PLA compositions (L₆-F₃₀-L₆ and L₁₆-F₃₀-L₁₆) where a
315 significant loss in the Young's modulus is observed after the first elongation to 20% strain (Figure
316 10). Increased PLA compositions reduce the flexibility of the polymers and therefore its resistance
317 to permanent deformation. Finetuning of the PLA composition will enable a wide array of

318 mechanical properties with ranging degrees of elasticity. However, to accommodate a wide array
319 of applications, elastomers need to exhibit various degrees of toughness as some uses exert high
320 forces on the materials, for example windshields wipers, snowmobile tracks, etc. One way to
321 improve the overall toughness of the polymer samples described in this study would be to increase
322 the molecular weight of the poly(farnesene) midblock beyond 30 kg/mol. Another option would be
323 to functionalize side chains of the poly(farnesene)-diol via consecutive epoxidation and hydrolysis,
324 to yield hydroxyl groups along the backbone that can be used to initiate ROP of lactide. The
325 resulting bottlebrush polymers would provide additional mechanisms of tunability, as grafting
326 density and weight compositions can be modulated with respect to each other.²⁸ For the present
327 systems, however, it is worth noting that their distinct high T_g -low T_g -high T_g architecture may
328 enable alternative functionalities beyond bulk mechanical reinforcement. In particular, this
329 architecture could promote adhesive behavior, as the combination of soft, low- T_g domains with
330 rigid, high- T_g end blocks is known to induce tack.⁷⁵ We therefore propose that future studies
331 assess the adhesive properties of these materials, which may open pathways toward their use in
332 pressure-sensitive adhesives or related soft-matter applications. Additionally, the block
333 copolymers described here may be suitable as compatibilizers for existing-, or novel polymer
334 blends in an effort to improve mechanical properties.

335

336



337

338 **Figure 10:** Cyclic stress strain curves for high molecular weight LFL block copolymers

339

340

341 **Quantifying waste generation in LFL block copolymer synthesis: an E factor perspective.**

342 With the ever-growing demand for society to transition to circular economies, renewables and bio-

343 based resources play an important role. Waste generation in chemical manufacturing is among

344 the biggest contributors to adverse environmental effects. This led to a paradigm shift in how the

345 efficiency of chemical processes is assessed.⁷⁶ The E (environmental) factor, introduced by Roger

346 Sheldon in the 1990s, brought attention to the amount of non-product material, such as solvents,

347 by-products and reagents, that is generated using the formula:

$$348 \quad E \text{ factor} = \frac{\text{weight of raw materials} - \text{weight of desired product}}{\text{weight of desired product}}$$

349 In doing so, value is given to the waste of a chemical process as opposed to the more traditional

350 *atom economy* metric, whereby efficiency is determined based only on the molecular weights of

351 the initial reagents and products. A lower E-factor suggests a lower environmental impact, and

352 years of its implementation in the chemical industry has given valuable insight as to how the E-

353 factor compares for various processes (Table 3).

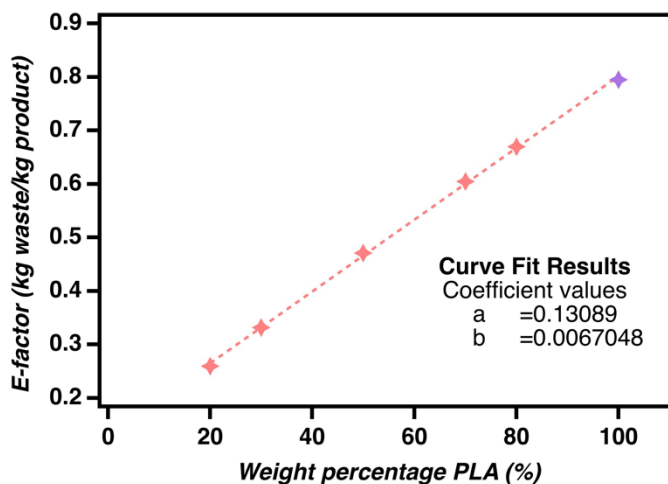
354

355 **Table 3:** A summary of E-factors within the chemical industry. Adapted from R. A. Sheldon,
356 *Green Chemistry*, 2023, 25, 1704-1728

Industry	Product tonnage (p/a)	E-factor (kgs/kg product)
Oil refining	10^6 - 10^8	<0.1
Bulk chemicals	10^4 - 10^6	<1-5
Fine chemicals	10^2 - 10^4	5-50
Pharmaceuticals	10 - 10^3	25-100

357
358 Using the above formula, an E-factor was calculated for LFL block copolymers and compared to
359 the synthesis of PLA homopolymer using an analogous synthesis method (i.e. in batch).
360 Assuming 90% yield and 90% solvent recovery, E-factor ranged between 0.26 and 0.67,
361 depending most critically on the weight composition of PLA in the final construct. The biggest
362 determining factor is the concentration of lactide monomer (relative to solvent) in the initial reagent
363 mixture, with higher concentrations resulting in less solvent loss per kg of product formed. For the
364 same reason, higher PLA weight compositions require more solvent per kg product due to the
365 increased lactide monomer. When compared to the synthesis of PLA homopolymer, block
366 copolymer synthesis is more efficient per kg product formed, largely because solvent is
367 substituted with a component that ultimately is incorporated into the final product. (Table S2),
368 (Figure 11).

L-F-L block copolymer synthesis



369
 370 **Figure 11:** Effect of increasing the PLA weight percentage on the E-factor in LFL block copolymer
 371 synthesis. For reference, the synthesis of PLA homopolymer is indicated in purple on the graph.

372
 373 The polynomial fit used to quantify the effect of PLA molecular weight in homopolymer synthesis
 374 suggests that from low to moderate degrees of polymerization (50 – 150), the E-factor decreases
 375 as product mass increases faster than waste generation. The quadratic term eventually catches
 376 up, with the E-factor plateauing as reaction efficiency starts to diminish. Additionally, extrapolation
 377 of the data gives us the most efficient degree of polymerization by taking the derivative:

378
$$\frac{dE}{dDP} = -0.00016558 + 2 * 3.3306 * 10^{-7} * DP$$

379 and setting it to zero:

380
$$0 = -0.00016558 + 6.6612 * 10^{-7} * DP$$

381 To solve for DP:

382
$$DP = \frac{0.00016558}{(6.6612 * 10^{-7})} = 248.6$$

383 Using these data extrapolations, experimental set-ups can be modulated to optimize efficiency.

384

385 Comparing the effect of weight fraction PLA incorporated in the LFL block copolymers provides
386 additional insight into the ROTEP of lactide. As shown in Figure 11, increasing the PLA weight
387 fraction results in a linearly increasing E-factor, described by the equation:

$$388 \quad E(x) = 0.12604 + 0.0068271 * x$$

389 where x represents the PLA weight percentage, and E(x) is the corresponding E-factor. This linear
390 relationship indicates that for every 1% increase in PLA content, the E-factor increases by
391 approximately 0.00683 units. Assuming a final product mass of 1 gram, this corresponds to an
392 increase in waste of roughly 6.83 mg per 1% increase in PLA. This emphasizes how process
393 design decisions—such as maintaining a constant lactide concentration across samples—can
394 significantly affect green metrics. It demonstrates that ROTEP of lactide does not become
395 necessarily less sustainable as weight fractions are increased, it's that the conditions needed to
396 scale its contribution in the block copolymer increase the amount of starting materials (i.e.
397 solvent).

398 Analyses like the one performed in this study tell us that E-factor metrics are not just
399 chemistry dependent; they are also process dependent and are crucial when optimizing reactions
400 when sustainability and eventual scale-up is the goal. Employing continuous flow production
401 protocols in the preparation of complex block copolymers can offer significant advantages in terms
402 of waste minimization and efficiency.

403 Importantly, the E-factor calculations employed here neglect the process required to
404 manufacture the poly(farnesene) mid-blocks. However, this process is already conducted on a
405 commercial scale, and most likely has a minimal impact on the total E-factor. It should be noted
406 however, that in terms of biodegradability, the incorporation of farnesene segments alters the
407 degradation pathway. Specifically, the PLA blocks, which contain hydrolytically labile ester bonds,
408 are expected to degrade readily under composting conditions. In contrast, the farnesene domains,
409 being hydrocarbon-based and non-hydrolyzable, are more resistant and will likely degrade more
410 slowly.⁷⁷ This reduced ester content, and the increased hydrophobicity introduced by farnesene

411 are likely to lower the overall degradation rate relative to pure PLA. Nevertheless, as farnesene
412 is a bio-based terpene derivative, we anticipate that it will undergo slow oxidative and microbial
413 breakdown over extended timescales.⁷⁸ However, the exact biodegradation pathways for the
414 polyfarnesene materials requires a closer evaluation to determine concretely.

415

416 **CONCLUSION**

417 We have synthesized a variety of PL-PF-PL block copolymers with varying PLA weight
418 compositions and PF molecular weight and demonstrated an intricate relationship between the
419 two parameters using numerous analytical techniques. Using ¹H-NMR and SEC analysis we
420 verified the incorporation of a target PLA composition with a high degree of precision. These
421 results allude to a change in microphase separation when going from low- to high molecular
422 weight block copolymers, which was confirmed using SAXS analysis and TEM imaging. Low
423 molecular weight block copolymers self-assemble into lamellar microdomains, with the lamellar
424 thickness being proportional to the weight composition of PLA, whereas high molecular weight
425 samples tend to form hexagonally packed cylinders of PLA in a matrix of poly(farnesene), with
426 cylinder diameter and inter-cylinder distances being proportional to the PLA composition.
427 Additionally, we demonstrated that in order to obtain elastomeric properties, the LFL block
428 copolymers should have a high molecular weight (>30 kg/mol) as the periodicity in the cylindrical
429 microdomains efficiently store potential energy, leading to typical elastomer-like mechanical
430 properties at low PLA weight compositions. These mechanical properties were then investigated
431 using a tensile tester, where it was clear that even a small compositional change (L₃-F₃₀-L₃ to L₆-
432 F₃₀-L₆) can have dramatic influence on the elastomeric behavior. Increasing PLA compositions
433 even further (L₂₇-F₃₀-L₂₇ and L₆₀-F₃₀-L₆₀) will drastically increase the Young's modulus but
434 significantly compromise flexibility. As for lower molecular weight block copolymers, no real
435 elastomeric effect is present across the spectrum of PLA weight compositions, with low

436 compositions behaving like a viscous fluid which can potentially be studied for adhesive properties
437 in a follow up study.

438 Overall, this study provides us with a very clear overview of the mechanical-, thermal-,
439 and morphological changes that occur when changing the molecular weight, and the weight
440 composition of the hard block in a fully renewable ABA block copolymer system. These results
441 provide a basic guideline of what polymer composition to target in order to achieve specific
442 material properties for various applications, in addition to some detailed insight into the molecular
443 mechanisms at play that lead to the broad spectrum of polymer properties. Notably, the highest
444 molar mass samples exhibit material properties that fall far short of other PLA-containing triblock
445 copolymers. We surmise that this may be a function of the molecular architecture of the
446 polyfarnesene block and we are keen to further explore the consequences of conformation on
447 physical attributes for these fully renewable thermoplastics. Finally, a study into the E-factor
448 analysis of LFL block copolymer synthesis compared to PLA homopolymer synthesis prepared
449 using an analogous reaction protocol highlights the effect of experimental conditions in the overall
450 sustainability of various materials, as well as steps that can be taken to minimize the overall E-
451 factor.

452 **ASSOCIATED CONTENT**

453 **Supporting Information**

454 The Supporting Information is available free of charge at:

455 *Additional experimental details, materials, and methods, including ¹H NMR spectra, GPC*
456 *chromatograms, DSC thermograms, and TEM micrographs.*

457

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473

474 **Author Contributions**

475 The manuscript was written through contributions from all authors. All authors have approved the
476 final version of the manuscript.

477 **Funding**

478 The authors are grateful for funding from the Bijzonder Onderzoeks Fonds (BOF) scheme under
479 contract BOF22KP04 and from the Flemish Research Foundation (FWO) under contract
480 G092023N and contract 1S19023N. Additionally we acknowledge the BM26 (DUBBLE) Beamline
481 at the European Synchrotron Radiation Facility (ESRF) for provision of synchrotron radiation
482 facilities under proposal number A26-2-984.

483 **Notes**

484 The authors declare no conflicts.

485

486 **ACKNOWLEDGMENTS**

487 The authors are grateful for helpful assistance from dr. Martin Rosenthal from the BM26
488 (DUBBLE) Beamline at the European Synchrotron Radiation Facility (ESRF).

489

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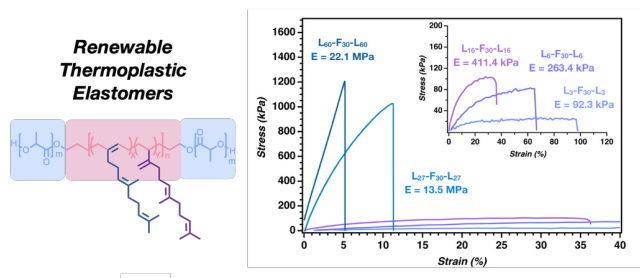
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703 **For Table of Contents**



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706 Renewable block polymers have been synthesized in a process with improved sustainability,
707 shown to self-assemble and exhibit elastomeric behavior.

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